

Office europa des brevets

(12)

EUROPEAN PATENT APPLICATION

(43) Date of publication:

08.05.1996 Bulletin 1996/19

(21) Application number: 95302225.8

(22) Date of filing: 03.04.1995

(84) Designated Contracting States: DE ES FR GB

(30) Priority: 02.11.1994 JP 293880/94 21.02.1995 JP 56691/95

(71) Applicant: MITSUI MINING & SMELTING CO., LTD. Tokyo (JP)

(51) Int. Cl.6: B23K 35/26

// C22C13/00

(72) Inventors:

· Ninomiya, Ryuji Ageo-shi, Saitama-ken 362 (JP)

· Matsunaga, Junichi Ageo-shi, Saitama-ken 362 (JP)

(74) Representative: Perry, Robert Edward **GILL JENNINGS & EVERY Broadgate House** 7 Eldon Street London EC2M 7LH (GB)

Lead-free solder (54)

A lead-free solder consisting of 1 to 4 wt% of Ag, either one or both of Bi and In of their respective amounts meeting the conditions that the value of expression (1), A, is equal to or greater than 5.00 and that the value of expression (2), B, is equal to or less than 6.90, and Sn for the balance:

$$A = [Ag wt\%] + 1.23 [Bi wt\%] + 0.52 [ln wt\%]$$
 (1)

$$B = [Ag wt\%] + 1.19 [Bi wt\%] + 0.50 [ln wt\%]$$
 (2)

This solder has a tensile strength and an elongation as high as those of conventional Pb-Sn solder without containing neither lead nor cadmium which can cause environmental contamination.

Description

BACKGROUND OF THE INVENTION

a) Field of the Invention

This invention relates to a lead-free solder which has mechanical properties as good as those of conventional Pb-Sn solder.

b) Description of the Prior Art

Pb-Sn alloy of eutectic or near-eutectic composition is well known as a representative solder. Zn-Cd alloy, which has a greater strength than Pb-Sn eutectic solder, is also known. Amid the increased concern with environmental protection in recent years, the harmfulness of lead and the noxious effect of vaporized cadmium on workers have become a problem with the former and latter solders, respectively, and a solder replacing them is needed.

To meet this need, zinc or tin base solders which contain neither harmful lead nor cadmium were proposed. The solders hitherto proposed, however, do not have the mechanical properties as good as conventional Pb-Sn solder. 3.5Ag-solder, for example, has a problem of a low tensile strength, though it has a satisfactory elongation. In this present situation, a solder which does not contain harmful lead or cadmium and has both a satisfactory high tensile strength and a satisfactory high elongation.

SUMMARY OF THE INVENTION

20

25

30

50

55

The object of the present invention is to provide a solder which does not contain harmful elements such as lead and cadmium and still has both a tensile strength and an elongation as high as those of Pb-Sn solder.

The solder of the present invention consists of 1 to 4 wt% of Ag, either one or both of Bi and In of their respective amounts meeting the conditions that the value of expression (1), A, is equal to or greater than 5.00 and that the value of expression (2), B, is equal to or less than 6.90, and Sn for the balance.

$$A = [Ag wt\%] + 1.23 [Bi wt\%] + 0.52 [In wt\%]$$
 (1)

$$B = [Ag wt\%] + 1.19 [Bi wt\%] + 0.50 [in wt\%]$$
 (2)

By this composition, the solder of the present invention has a tensile strength equal to or greater than 4.0 kgf/mm² and an elongation equal to or greater than 30% which are mechanical properties as good as those of conventional Pb-Sn solder.

BRIEF DESCRIPTION OF THE DRAWINGS

Figure 1 shows the relationship between the amounts of In and Bi to be added when the Ag content is 1 wt% (Sn for the balance).

Figure 2 shows the relationship between the amounts of In and Bi to be added when the Ag content is 2 wt% (Sn for the balance Sn).

Figure 3 shows the relationship between the amounts of In and Bi to be added when the Ag content is 3 wt% (Sn for the balance).

Figure 4 shows the relationship between the amounts of In and Bi to be added when the Ag content is 4 wt% (Sn for the balance).

DESCRIPTION OF THE PREFERRED EMBODIMENTS

In the present invention, silver has an effects of improving the heat resistance, lowering the fusing temperature, and increasing the strength, spreadability and brightness of the solder by a little amount contained. The silver content appropriate for this purpose is equal to or greater than 1 wt%. The upper limit of the silver content is 4 wt%, taking into account the expensiveness of silver.

The Bi and In contents are within the ranges such that the combination of their contents fall in the hatched area in Figures 1 to 4 for 1 wt%, 2 wt%, 3 wt% and 4 wt%, respectively, which can be defined by the conditions that the value of expression (1), A, is equal to or greater than 5.00 and that the value of expression (2), B, is equal to or less than 6.90, where expressions (1) and (2) represent the upper and lower boundary lines of the hatched area in Figure 3.

the value of expression (2), B, is greater than 6.90, on the other hand, the elongation of the solder is lower than 30%.

Since the solder of the presentation may contain either one or both of Bi In of the amounts as described above, it may be Sn-Ag-Bi or Sn-Ag-In ternary alloy or Sn-Ag-Bi-In quarternary alloy.

When the solder is Sn-Ag-Bi ternary alloy without In, expressions (1) and (2) become as follows:

$$A = [Ag wt\%] + 1.23 [Bi wt\%]$$
 (1')

$$B = [Ag \text{ wt\%}] + 1.19 [Bi \text{ wt\%}]$$
 (2')

From these expressions and the aforementioned range of Ag, the Bi content is greater than 0.81 wt% and smaller than 6.4 w%.

If the Bi content is smaller than 0.81 wt%, the tensile strength of the solder is lower than 4.0 kgf/mm². If greater than 6.4 wt%, the elongation of the solder is lower than 30%.

When the solder is Sn-Ag-In ternary alloy without Bi, expressions (1) and (2) become as follows:

$$A = [Ag wt\%] + 0.52 [in wt\%]$$
 (1")

$$B = [Ag wt\%] + 0.50 [In wt\%]$$
 (2")

From these expressions and the aforementioned range of Ag, the In content is greater than 1.9 wt% and smaller than 12 wt%.

If the In content is smaller than 1.9 wt%, the tensile strength of the solder is lower than 4.0 kgf/mm². If greater than 12 wt%, the elongation of the solder is lower than 30%. However, it is preferable to set the upper limit of the In content to 4.0 wt% because In is expensive.

When the solder is Sn-Ag-Bi-In quarternary alloy, the Bi content is within the range of 0.81 to 5.0 wt% and the In content is within the range of 1.9 to 12 wt% from expressions (1) and (2) and the range of the Ag content.

Examples

i,

5

10

15

20

30

40

45

50

55

Sn, Ag, Bi, In and Pb were weighed out to prepare compositions listed in Table 1, each composition being 10 kg in total. These compositions were melted at 300 °C in an electric furnace in the air using a graphite crucilbe. After all metals had melted, the molten metals were thoroughly stirred to prevent the segregation caused by the gravity and cast into solder pieces of 10 mm in thickness using a mold of 150 mm x 60 mm with the height of 150 mm in inner dimensions. Test samples were cut from the bottom part of the solder pieces according to the JIS 4 standards with a machine, and their tensile strength and elongation were measured by the test method defined by the JIS Z2241 standards. The results are shown in Table 1. The strength and the elongation of Pb-Sn eutectic solder and 3.5Ag-Sn solder are also shown in

Table 1

	Chemi	cal Co	mposit	ion(wt%)	Tensile Strength	elongation
	Ag	Вi	In	Sn	(kgf/mm²)	(%)
	1	0	8	Bal.	4.1	35.9
	1	0	9.	Bal.	4.3	34.2
		0	10	Bal.	4.5	32.6
E	1	0	11	Bal.	4.7	31.0
۲	1	1	6	Bal.	4.1	35.3
a	1	1	7	Bal.	4.3	33.7
n (1	li	8	Bal.	4.5	32.0
,	1	1	9	Bal.	4.7	30.4
i	1	2	3	Bal.	4.0	36.4
9	ì	2 2 2 2 3 3		Bal.	4.2	34.8
- I	ī	2	4 5 6	Bal.	4.4	33.1
-	ī	2	6	Bal.	4.6	
ı	i	3	1	Bal.	4.1	31.5
	ì	3	2	Bal.	4.3	35.9
Ì	i	3	3		4.5 4.5	34.2
	1	3	4	Bal.	4.5	32.6
1	1	4	0	Bal.	4.7	30.9
	1	4		Bal.	4.3	33.7
- 1	1		1	Bal.	4.5	32.0
	1	4	2 6 7	Bal.	4.7	30.3
	2	0	b	Bal.	4.1	36.1
-	2	0	1	Bal.	4.2	34.4
	Z	0	8 9	Bal.	4.4	32.8
	2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	0	9	Bal.	4.6	31.2
- 1	2	1	4	Bal.	4.1	35.5
	2	1	5 6	Bal.	4.3	33.9
	2	1	6	Bal.	4.5	32.2
ı	2	1	7 2	Bal.	4.7	30.6
	2	1 2 2 2 2	2	Bal.	4.2	35.0
ı	2	2	3 4	Bal.	4.4	33.3
	2	2		Bal.	4.6	31.7
	2	2	5	Bal.	4.8	30.0
j	2	3	0 1	Bal.	4.2	34.4
	2	3	1	Bal.	4.4	32.8
- 1	2	3	2	Bal.	4.6	31.1
	2 2 3 3 3 3 3 3	3 4	2	Bal.	4.7	30.5
i	3	0		Bal.	4.0	36.2
	3	0	5	Bal.	4.2	34.6
	3	0	6	Bal.	4.4	33.0
	3	. 0	7	Bal.	4.6	31.4
- 1	3	1	2	Bal.	4.1	35.7
	3	ī	4 5 6 7 2 3	Bal.	4.3	34.1
	3 3 3 3 3	ī	4	Bal.	4.5	32.4
	3		5	Bal.	4.7	30.8
İ	3	2	4 5 0	Bal.	4.2	35.2
	3	2	1	Bal.	4.4	
	3	1 2 2 2		Bal.	4.6	33.5
	3	2	2 3	Bal.	4.0	31.9
				naT.	4./	30.2

	Chemi	cal Co	mposit	ion(wt%)	Tensile Strength	elongation
L	Ag	Bi	In	Sn	(kgf/mm²)	(%)
	3	3	0	Bal.	4.6	31.3
	4	0	2	Bal.	4.0	36.4
E	4	0	3 4	Bal.	4.2	34.8
x	4	0	4	Bal.	4.4	33.2
a	4	0		Bal.	4.6	31.5
m	4	1	0	Bal.	4.1	35.9
р	4	1 1	1	Bal.	4.3	34.2
1	4		2	Bal.	4.5	32.6
е	4	1	3	Bal.	4.7	31.0
	4	2	0	Bal.	4.5	32.1
	1	1 2 2	7	Bal.	4.7	30.4
İ	1			Bal.	3.9	37.5
C	1	0	12	Bal.	4.9	29.4
0	1	1	5	Bal.	3.9	37.0
m	1	1	10	Bal.	4.9	28,8
P	1	2	2 7	Bal.	3.8	38.1
a	1	1 2 2 3 3	7	Bal.	4.8	29.8
r	1	3	0	Bal.	3.9	37.5
a	1	3	5 3 5	Bal.	4.9	29.3
t	1 1	4	3	Bal.	4.9	28.7
li	2 2	0		Bal.	3.9	37.7
V	2		10	Bal.	4.8	29.5
е	2	1	3	Bal.	3.9	37.1
	2 2 2	1 2	8	Bal.	4.9	29.0
E	2	2		Bal.	3.9	36.6
x	2 2 . 2	2 3	6	Bal.	5.0	28.4
. a	2		3	Bal.	4.8	29.5
m		4	1	Bal.	4.9	28.9
P	3 3	0	3 8	Bal.	3.8	37.9
1		0	_	Bal.	4.8	29.7
е	3	1	1 6	Bal.	3.9	37.3
	3	1	6	Bal.	4.9	29.2
	3 3 3 3	2 3	4	Bal.	4.9	28.6
		1	1	Bal.	4.8	29.6
	4	. 0	1	Bal.	3.8	38.1
	4	0	6	Bal.	4.8	29.9
	4	1	4 2	Bal.	4.9	29.3
<u> </u>	4	2		Bal.	4.9	28.8
	P b -	63S	n		3.8	30.0
L	<u>Sn</u> -	3.5	Ag		2.0	70.0

Thus, by the present invention as described above, a solder with mechanical properties as good as those of conventional Pb-Sn solder is provided.

It is known from Table 1 that the solder of the present invention has a tensile strength equal to or greater than 4.0 kgf/mm² and an elongation equal to or greater than 30% which are satisfactory mechanical properties to attain the object of the present invention.

Jaims

1. A lead-free solder consisting of 1 to 4 wt% of Ag, either one or both of Bi and In of their respective amounts meeting the conditions that the value of expression (1), A, is equal to or greater than 5.00 and that the value of expression (2), B, is equal to or less than 6.90, and Sn for the balance:

$$A = [Ag wt\%] + 1.23 [Bi wt\%] + 0.52 [In wt\%]$$
 (1)

$$B = [Ag wt\%] + 1.19 [Bi wt\%] + 0.50 [In wt\%]$$
 (2)

2. The lead-free solder of claim 1 which is Sn-Ag-In ternary alloy with the In content smaller than 4 wt%.

FIG. 1

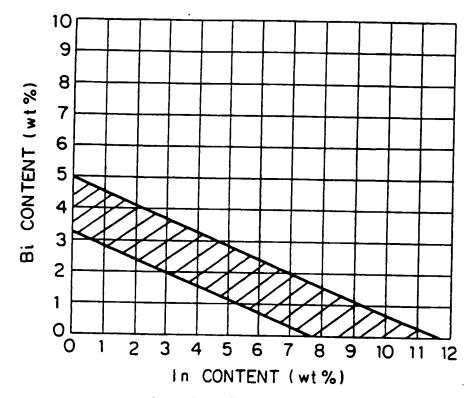
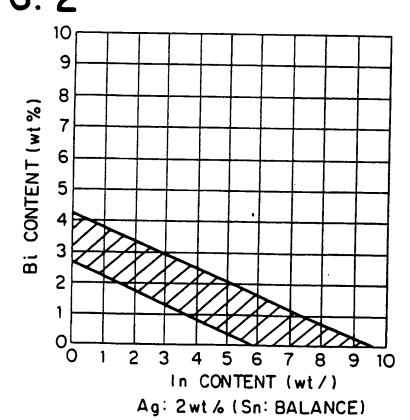


FIG. 2 Ag: 1 wt % (Sn: BALANCE)



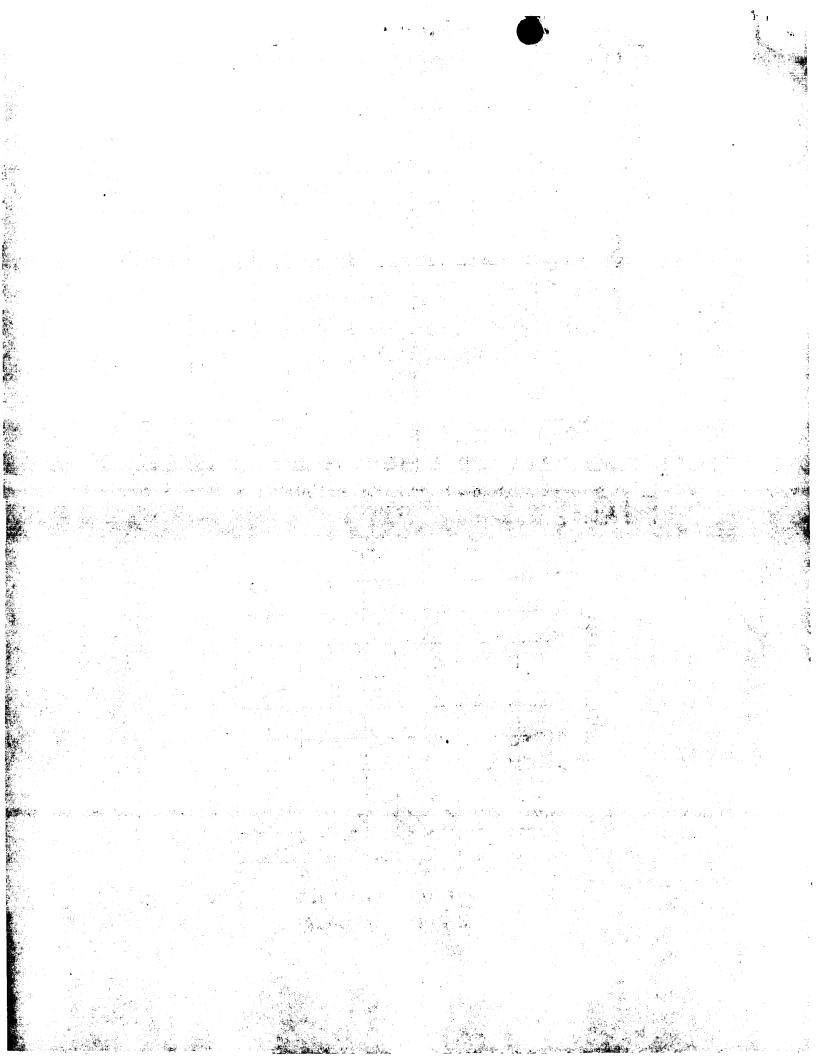
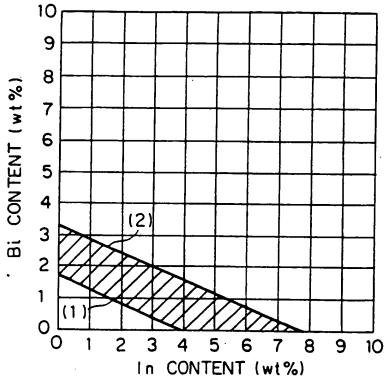
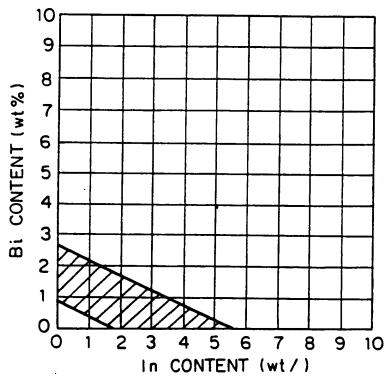


FIG. 3



Ag: 3wt% (Sn: BALANCE)

FIG. 4



Ag: 4wt % (Sn: BALANCE)

		Topology at
		The state of the s



EUROPEAN SEARCH REPORT

		IDERED TO BE RELEVAN	1	
Cutegory	Citation of document with of relevant p	indication, where appropriate,	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.CL6)
X	EP-A-0 568 952 (TH * claims 1,8; exam	E INDIUM CORPORATION) ple 1 *	1	B23K35/26 //C22C13/00
x	EP-A-0 499 452 (LA * claims 1,2; table	NCASHIRE FITTINGS)	1	
X	GB-A-131 299 (S.A. CABLERIE) * page 2, line 6 -		1	
į	PATENT ABSTRACTS OF vol. 17 no. 675 (M- & JP-A-05 228685 (September 1993, * abstract *	JAPAN -1526) ,13 December 1993 (MATSUSHITA ELECTRIC) 7	1	
\	US-A-5 328 660 (S.C	G. GONYA ET AL)		
l	EP-A-O 629 466 (INT MACHINES) * claims 1,2; examp	TERNATIONAL BUSINESS	1	TEGONG A SEP OS
ļ	Craims 1,2, Examp			TECHNICAL FIELDS SEARCHED (Int.Cl.6)
				·
	The present search report has be	en drawn up for all claims		
	Place of search THE HAGUE	Date of completion of the search		Examinar
X : partice Y : partice docum A : techno O : non-w	TEGORY OF CITED DOCUMEN tharly relevant if taken alone tharly relevant if combined with anot ent of the same category logical background ritten disclosure ediate document	E: earlier patent docus	underlying the intent, but publis the application other reasons	bed on, or

	- - -			· ens
	1	# # # # # # # # # # # # # # # # # # #	•	•
		13		
	And the second			
	the second			•
			in the second of the second o	•
			and the second s	
				•
				w · · ·
18 cm				
				•
			n de la Carlo de la Carlo de la Carlo de la Carlo de la Carlo de la Carlo de la Carlo de la Carlo de la Carlo d La Carlo de la	
			and the second of the second o	, · · · · · · · · · · · · · · · · · · ·
en en en en en en en en en en en en en e				
	The state of the s		The state of the s	
and the second second		en de la companya de la companya de la companya de la companya de la companya de la companya de la companya de La companya de la co		
			기계 교육 요시 그녀를 중 하는 것이라는 생활이 하다. 지원 기계 기계 기계 기계 기계 기계 기계 기계 기계 기계 기계 기계 기계	
				3.
		$\frac{\partial}{\partial x} = \frac{\partial}{\partial x} \left(\frac{\partial}{\partial x} - \frac{\partial}{\partial x} \right) = \frac{\partial}{\partial x} \left(\frac{\partial}{\partial x} - \frac{\partial}{\partial x} \right) = \frac{\partial}{\partial x} \left(\frac{\partial}{\partial x} - \frac{\partial}{\partial x} \right) = \frac{\partial}{\partial x} \left(\frac{\partial}{\partial x} - \frac{\partial}{\partial x} \right) = \frac{\partial}{\partial x} \left(\frac{\partial}{\partial x} - \frac{\partial}{\partial x} \right) = \frac{\partial}{\partial x} \left(\frac{\partial}{\partial x} - \frac{\partial}{\partial x} \right) = \frac{\partial}{\partial x} \left(\frac{\partial}{\partial x} - \frac{\partial}{\partial x} - \frac{\partial}{\partial x} \right) = \frac{\partial}{\partial x} \left(\frac{\partial}{\partial x} - \frac{\partial}{\partial x} - \frac{\partial}{\partial x} \right) = \frac{\partial}{\partial x} \left(\frac{\partial}{\partial x} - \frac{\partial}{\partial x} - \frac{\partial}{\partial x} \right) = \frac{\partial}{\partial x} \left(\frac{\partial}{\partial x} - \frac{\partial}{\partial x} - \frac{\partial}{\partial x} - \frac{\partial}{\partial x} \right) = \frac{\partial}{\partial x} \left(\frac{\partial}{\partial x} - \frac{\partial}{\partial x} - \frac{\partial}{\partial x} - \frac{\partial}{\partial x} \right) = \frac{\partial}{\partial x} \left(\frac{\partial}{\partial x} - \frac{\partial}{\partial $		
	A.,			
		(x,y) = (x,y) + (x,y)		ing the state of
		en en en en en en en en en en en en en e	en en en en en en en en en en en en en e	and the second s
Tagan da Maranga kada Maranga	The first of the first of the second of the	and the second of the second o	i ja kirika di di di di di di di di di di di di di	ing the second s
	Service Control of the			
	4. **	gan store sate .		Grand State of the
				B Charles
	and the second s			
	And the second s			
그 이 양성적인 및 4점 전환 42			garan da karantarian da karantarian da karantarian da karantarian da karantarian da karantarian da karantarian Manganan da karantarian da karantarian da karantarian da karantarian da karantarian da karantarian da karantar	
the second of th			and the second second	